

COMPONENTS:

1. Methane; CH₄; [74-82-8]
2. 2-Methylpropane (*isobutane*); C₄H₁₀; [75-28-5]

EVALUATOR:

Colin L. Young,
School of Chemistry,
University of Melbourne,
Parkville, Victoria 3052,
Australia.

March 1982

CRITICAL EVALUATION:

The most extensive sets of data on this system are those of Barsuk *et al.* (1). These data which cover the temperature range 198 to 377 K and are in reasonable agreement with those of Olds *et al.* (2) at 310.9 K and 344.25 K. There are significant discrepancies between the two sets of data at 377.6 K and near the critical region at the low temperatures.

References

1. Barsuk, S. D.; Skripka, V. G.; Benyaminovich, O. A.
Gazov. Prom. 1970, 15, 38.
2. Olds, R. H.; Sage, B. H.; Lacey, W. N.
Ind. Eng. Chem. 1942, 34, 1008.

COMPONENTS:			ORIGINAL MEASUREMENTS:							
1. Methane; CH ₄ ; [74-82-8]			Olds, R. H.; Sage, B. H.; Lacey, W. N. <i>Ind. Eng. Chem.</i> <u>1942, 34, 1008-1013.</u>							
2. 2-Methylpropane (<i>isobutane</i>); C ₄ H ₁₀ ; [75-28-5]										
VARIABLES:			PREPARED BY:							
C. L. Young										
EXPERIMENTAL VALUES:										
T/K (T/°F)	P/MPa	p/psi	Wt. fraction of methane in liquid, in vapor,	Mole fraction of methane in liquid, in vapor, x_{CH_4} y_{CH_4}						
310.9 (100)	0.55	80	0.00077	0.0248	0.00278	0.0843				
	0.69	100	0.00311	0.0929	0.01117	0.2704				
	1.03	150	0.00843	0.2049	0.02985	0.4826				
	1.38	200	0.01400	0.2869	0.04888	0.5929				
	2.07	300	0.02586	0.4031	0.08766	0.7097				
	2.76	400	0.03793	0.4646	0.1249	0.7585				
	3.45	500	0.05076	0.5076	0.1622	0.7886				
	4.14	600	0.06434	0.5380	0.1993	0.8082				
	4.83	700	0.07870	0.5594	0.2362	0.8213				
	5.52	800	0.09390	0.5740	0.2728	0.8298				
	6.21	900	0.1098	0.5826	0.4273	0.8348				
	6.89	1000	0.1266	0.5859	0.3441	0.8366				
	7.58	1100	0.1449	0.5845	0.3802	0.8358				
	8.27	1200	0.1648	0.5785	0.4166	0.8324				
	8.96	1300	0.1870	0.5673	0.4543	0.8259				
	9.65	1400	0.2123	0.5498	0.4938	0.8155				
	10.34	1500	0.2430	0.5233	0.5374	0.7989				
	11.03	1600	0.2858	0.4810	0.5916	0.7703				
	11.58	1679	0.3800	0.3800	0.6893	0.6893				
344.3 (160)	1.38	200	0.00359	0.0492	0.01287	0.1577				
	2.07	300	0.01345	0.1468	0.04702	0.3838				
(cont.)										
AUXILIARY INFORMATION										
METHOD/APPARATUS/PROCEDURE:			SOURCE AND PURITY OF MATERIALS:							
PVT cell charged with mixture of known composition. Pressure measured with pressure balance. Temperature measured with resistance thermometer. Bubble point and dew point determined for various compositions from discontinuity in PV isotherm. Coexisting liquid and gas phase properties determined by graphical means. Details in ref. (1).			1. Crude sample treated for removal of higher alkanes, carbon dioxide and water vapor. Final purity 99.9 mole per cent. 2. Phillips Petroleum sample, purity at least 99.97 mole per cent.							
			ESTIMATED ERROR: $\delta T/K = \pm 0.1$; $\delta P/MPa = \pm 0.007$; $\delta x_{CH_4} = \pm 0.001$; $\delta y_{CH_4} = \pm 0.005$ (estimated by compiler).							
			REFERENCES:							
			1. Sage, B. H.; Lacey, W. N. <i>Trans. Am. Inst. Mining Met. Engrs.</i> <u>1940, 136, 136.</u>							

COMPONENTS:			ORIGINAL MEASUREMENTS:			
1. Methane; CH ₄ ; [74-82-8]			Olds, R. H.; Sage, B. H.; Lacey, W. N.			
2. 2-Methylpropane (<i>isobutane</i>) C ₄ H ₁₀ ; [75-28-5]			<i>Ind. Eng. Chem.</i> <u>1942</u> , 34, 1008-1013.			
EXPERIMENTAL VALUES:						
T/K (T/°F)		P/MPa p/psi	Wt. fraction of methane in liquid, in vapor,	Mole fraction of methane <i>x</i> _{CH₄} <i>y</i> _{CH₄}		
344.3 (160)		2.76 400	0.02381 0.2139	0.08112 0.4962		
3.45		500	0.03481 0.2633	0.1155 0.5640		
4.14		600	0.04645 0.2998	0.1499 0.6078		
4.83		700	0.05875 0.3262	0.1843 0.6367		
5.52		800	0.07200 0.3454	0.2192 0.6563		
6.21		900	0.08624 0.3587	0.2546 0.6694		
6.89		1000	0.1020 0.3677	0.2913 0.6779		
7.58		1100	0.1189 0.3683	0.3281 0.6785		
8.27		1200	0.1381 0.3631	0.3671 0.6736		
8.96		1300	0.1623 0.3477	0.4122 0.6586		
9.65		1400	0.1980 0.3156	0.4719 0.6253		
10.05		1457	0.2580 0.2580	0.5572 0.5572		
377.6 (220)		2.76 400	0.00871 0.0472	0.03082 0.1520		
3.45		500	0.01912 0.0876	0.06590 0.2579		
4.14		600	0.03030 0.1182	0.1016 0.3267		
4.83		700	0.04160 0.1387	0.1358 0.3682		
5.52		800	0.05460 0.1521	0.1729 0.3937		
6.21		900	0.07030 0.1580	0.2149 0.4045		
6.89		1000	0.09370 0.1488	0.2723 0.3875		
7.14		1035	0.1230 0.1230	0.3367 0.3367		

COMPONENTS: 1. Methane; CH ₄ ; [74-82-8] 2. 2-Methylpropane; C ₄ H ₁₀ ; [75-28-5]		ORIGINAL MEASUREMENTS: Barsuk, S.D.; Skripka, V.G.; Benyaminovich, O.A. <i>Gazov. Prom.</i> 1970, 15, 38-41.		
VARIABLES: Temperature, pressure		PREPARED BY: C.L. Young		
EXPERIMENTAL VALUES:				
T/K	P/10 ⁵ Pa	Mole fraction of methane in liquid, x_{CH_4}	Mole fraction of methane in vapor, y_{CH_4}	
198.15	4.9 9.8 19.6 29.4 39.2 49.0 53.4	0.092 0.184 0.354 0.523 0.698 0.826 0.981	0.993 0.996 0.998 0.998 0.997 0.995 0.981	
213.15	4.9 9.8 19.6 29.4 39.2 49.0 58.8 68.6 70.6	0.074 0.141 0.269 0.394 0.516 0.635 0.755 0.890 0.960	0.981 0.988 0.993 0.996 0.995 0.993 0.991 0.973 0.960	
233.15	4.9 9.8 19.6 29.4 39.4 49.0 58.8	0.058 0.111 0.201 0.293 0.382 0.472 0.562	0.948 0.966 0.980 0.985 0.986 0.985 0.983	
AUXILIARY INFORMATION				
METHOD/APPARATUS/PROCEDURE: Recirculating vapor flow apparatus fitted with magnetic stirrer. Temperature measured with platinum resistance thermometer. Liquid and gas phases analysed by gas chromatography. Details in source and ref. (1).	SOURCE AND PURITY OF MATERIALS: Both samples had purity of 99.5 mole per cent.			
ESTIMATED ERROR: $\delta T/K = \pm 0.1$; $\delta P/10^5 Pa = \pm 0.4$; $\delta x_{CH_4}, \delta y_{CH_4} = \pm 3\%$				
REFERENCES: 1. Skripka, V.G.; Barsuk, S.D.; Nikitina, I.E.; Benyaminovich, O.A. <i>Gazov. Prom.</i> 1964, 14, 41.				

COMPONENTS:

ORIGINAL MEASUREMENTS:

1. Methane; CH₄; [74-82-8]
 2. 2-Methylpropane; C₄H₁₀; [75-28-5]

Barsuk, S.D.; Skripka, V.G.;
 Benyaminovich, O.A.
Gazov. Prom. 1970, 15, 38-41.

EXPERIMENTAL VALUES:

T/K	P/10 ⁵ Pa	Mole fraction of methane	
		x _{CH₄}	y _{CH₄}
233.15	68.6	0.651	0.978
	78.5	0.745	0.971
	88.3	0.850	0.952
	91.2	0.908	0.908
253.15	4.9	0.038	0.670
	9.8	0.080	0.894
	19.6	0.155	0.946
	29.4	0.230	0.963
	39.2	0.306	0.967
	49.0	0.377	0.966
	58.8	0.449	0.964
	68.6	0.520	0.962
	78.5	0.593	0.957
	88.3	0.670	0.948
	98.1	0.754	0.924
	106.5	0.856	0.856
273.15	4.9	0.024	-
	9.8	0.060	0.829
	19.6	0.132	0.899
	29.4	0.200	0.921
	39.2	0.268	0.930
	49.0	0.331	0.935
	58.8	0.393	0.938
	68.6	0.457	0.939
	78.5	0.521	0.937
	88.3	0.586	0.931
	98.1	0.651	0.918
	107.9	0.714	0.889
	114.9	0.810	0.810
293.15	4.9	0.012	0.340
	9.8	0.043	0.741
	19.6	0.102	0.827
	29.4	0.163	0.860
	39.2	0.222	0.880
	49.0	0.281	0.890
	58.8	0.339	0.898
	68.6	0.396	0.899
	78.5	0.454	0.896
	88.3	0.512	0.889
	98.1	0.574	0.877
	107.9	0.638	0.854
	117.7	0.725	0.778
	117.9	0.750	0.750
310.95	4.9	0.002	0.030
	9.8	0.030	0.485
	19.6	0.084	0.707
	29.4	0.140	0.770
	39.2	0.194	0.805
	49.0	0.249	0.824
	58.8	0.303	0.834
	68.6	0.357	0.835
	78.5	0.412	0.833
	88.3	0.467	0.824
	98.1	0.524	0.807
	107.9	0.603	0.760
	112.0	0.690	0.690

COMPONENTS:

1. Methane; CH₄; [74-82-8]
 2. 2-Methylpropane; C₄H₁₀; [75-28-5]

ORIGINAL MEASUREMENTS:

Barsuk, S.D.; Skripka, V.G.;
 Benyaminovich, O.A.
Gazov. Prom. 1970, 15, 38-41.

EXPERIMENTAL VALUES:

T/K	P/10 ⁵ Pa	Mole fraction of methane in liquid, <i>x</i> _{CH₄}	Mole fraction of methane in vapor, <i>y</i> _{CH₄}
344.25	14.7	0.018	0.231
	19.6	0.045	0.377
	29.4	0.096	0.531
	39.2	0.145	0.605
	49.0	0.196	0.645
	58.8	0.248	0.670
	68.6	0.301	0.680
	78.5	0.358	0.676
	88.3	0.425	0.654
	97.0	0.558	0.558
377.55	24.5	0.018	0.111
	29.4	0.044	0.200
	39.2	0.096	0.306
	49.0	0.150	0.377
	58.8	0.208	0.405
	68.6	0.316	0.358
	68.9	0.337	0.337